

Ethyl 4-[[3-(adamantan-1-yl)-4-phenyl-5-sulfanylidene-4,5-dihydro-1H-1,2,4-triazol-1-yl]methyl]piperazine-1-carboxylate

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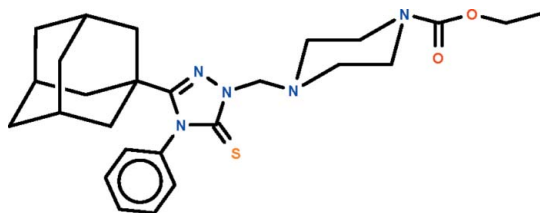
Received 19 December 2011; accepted 26 December 2011

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.061; wR factor = 0.156; data-to-parameter ratio = 18.4.

The title molecule, $\text{C}_{26}\text{H}_{35}\text{N}_5\text{O}_2\text{S}$, displays a chair-shaped piperazine ring, as well as a planar triazole ring whose phenyl substituent is perpendicular to the mean plane of the five-membered ring [dihedral angle = 90.00 (13)°]. The methylene substituent on the piperazine ring occupies an equatorial site. Weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding is present in the crystal structure. The crystal studied was a non-merohedral twin, with a 33.9 (3)% minor component.

Related literature

For the synthesis and application of the title compound, see: El-Emam & Ibrahim (1991). For the separation of non-merohedrally twinned diffraction data, see: Spek (2009).



Experimental

Crystal data

$\text{C}_{26}\text{H}_{35}\text{N}_5\text{O}_2\text{S}$
 $M_r = 481.65$
 Monoclinic, $P2_1/c$
 $a = 12.0469$ (6) Å
 $b = 20.9213$ (10) Å
 $c = 10.3249$ (5) Å
 $\beta = 109.851$ (6)°
 $V = 2447.6$ (2) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.17$ mm⁻¹
 $T = 100$ K
 $0.30 \times 0.20 \times 0.10$ mm

Data collection

Agilent SuperNova Dual diffractometer with an Atlas detector
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2010)
 $T_{\min} = 0.952$, $T_{\max} = 0.984$
 17138 measured reflections
 5655 independent reflections
 4588 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.061$
 $wR(F^2) = 0.156$
 $S = 1.11$
 5655 reflections
 308 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.92$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.40$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C6}-\text{H6}\cdots\text{O2}^i$	0.95	2.48	3.406 (3)	166
$\text{C8}-\text{H8}\cdots\text{O2}^{ii}$	0.95	2.46	3.207 (3)	135

Symmetry codes: (i) $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x + 1, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Agilent, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

We thank the Deanship of Scientific Research and the Research Center of the College of Pharmacy, King Saud University, and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5418).

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supplementary materials

Acta Cryst. (2012). E68, o531 [doi:10.1107/S1600536811055668]

Ethyl 4-[[3-(adamantan-1-yl)-4-phenyl-5-sulfanylidene-4,5-dihydro-1*H*-1,2,4-triazol-1-yl]methyl]piperazine-1-carboxylate

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Comment

We reported the the synthesis, anti-inflammatory and analgesic properties of 3-(1-adamantyl)-4-substituted-5-mercapto-1,2,4-triazole derivatives (El-Emam & Ibrahim, 1991). The triazole ring, which possesses a secondary nitrogen site next to a double-bond sulfur, is capable of undergoing a Mannich reaction with an *N*-substituted piperazine derivative to yield a new class of chemotherapeutic compounds. The C₂₆H₃₅N₅O₂S molecule (Scheme I, Fig. 1) displays a chair-shaped piperazine ring, as well as a planar triazole ring whose phenyl substituent is perpendicular to the mean plane of the five-membered ring (dihedral angle 90.0 (1) °).

Experimental

5-(1-Adamantyl)-4-phenyl-1,2,4-triazole-3-thiol was synthesized according to a reported procedure (El-Emam & Ibrahim, 1991). The compound (2 mmol), ethyl piperazine-1-carboxylate (2 mmol) and a 37% formaldehyde solution (0.5 ml) in ethanol (8 ml), was heated for 15 minutes. Stirring was continued for 12 h at room temperature. The product was filtered, washed with water, dried, and recrystallized from ethanol to yield (80%) of the title compound as colorless crystals, m.p. 465–467 K.

Refinement

Carbon-bound H-atoms were placed in calculated positions [C–H 0.95 to 1.00 Å, $U_{\text{iso}}(\text{H})$ 1.2 to 1.5 $U_{\text{eq}}(\text{C})$] and were included in the refinement in the riding model approximation.

The crystal studied is a non-merohedral twin. The twin components were separated by using *PLATON* (Spek, 2009).

Figures

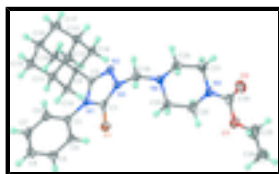


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of C₂₆H₃₅N₅O₂S at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

Ethyl 4-[[3-(adamantan-1-yl)-4-phenyl-5-sulfanylidene-4,5-dihydro-1*H*-1,2,4-triazol-1-yl]methyl]piperazine-1-carboxylate

Crystal data

C₂₆H₃₅N₅O₂S

$F(000) = 1032$

supplementary materials

$M_r = 481.65$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 12.0469$ (6) Å

$b = 20.9213$ (10) Å

$c = 10.3249$ (5) Å

$\beta = 109.851$ (6)°

$V = 2447.6$ (2) Å³

$Z = 4$

$D_x = 1.307$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5944 reflections

$\theta = 2.3$ – 27.5 °

$\mu = 0.17$ mm⁻¹

$T = 100$ K

Prism, colorless

$0.30 \times 0.20 \times 0.10$ mm

Data collection

Agilent SuperNova Dual
diffractometer with an Atlas detector

Radiation source: SuperNova (Mo) X-ray Source

Mirror

Detector resolution: 10.4041 pixels mm⁻¹

ω scan

Absorption correction: multi-scan
(CrysAlis PRO; Agilent, 2010)

$T_{\min} = 0.952$, $T_{\max} = 0.984$

17138 measured reflections

5655 independent reflections

4588 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.046$

$\theta_{\max} = 27.6$ °, $\theta_{\min} = 2.3$ °

$h = -15 \rightarrow 15$

$k = -27 \rightarrow 27$

$l = -10 \rightarrow 13$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.061$

$wR(F^2) = 0.156$

$S = 1.11$

5655 reflections

308 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0559P)^2 + 1.7988P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.92$ e Å⁻³

$\Delta\rho_{\min} = -0.40$ e Å⁻³

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	1.07961 (6)	0.35628 (3)	0.77520 (7)	0.02400 (17)
N2	1.06373 (17)	0.35443 (10)	0.3905 (2)	0.0199 (4)
N1	1.14387 (16)	0.41508 (9)	0.5741 (2)	0.0162 (4)
N3	1.03774 (17)	0.33163 (10)	0.5027 (2)	0.0191 (4)
N4	0.85438 (18)	0.27336 (10)	0.4214 (2)	0.0221 (5)
N5	0.61492 (19)	0.24428 (11)	0.3815 (3)	0.0304 (5)
O1	0.47688 (16)	0.25081 (9)	0.4818 (2)	0.0302 (4)

O2	0.50958 (18)	0.15641 (9)	0.3949 (2)	0.0334 (5)
C1	1.0859 (2)	0.36674 (11)	0.6181 (3)	0.0185 (5)
C2	1.12871 (19)	0.40508 (11)	0.4359 (2)	0.0161 (5)
C3	1.2096 (2)	0.46416 (11)	0.6668 (2)	0.0158 (5)
C4	1.1530 (2)	0.52024 (12)	0.6799 (3)	0.0197 (5)
H4	1.0716	0.5262	0.6288	0.024*
C5	1.2166 (2)	0.56774 (13)	0.7686 (3)	0.0243 (6)
H5	1.1790	0.6065	0.7784	0.029*
C6	1.3355 (2)	0.55824 (14)	0.8428 (3)	0.0263 (6)
H6	1.3796	0.5906	0.9029	0.032*
C7	1.3895 (2)	0.50143 (14)	0.8290 (3)	0.0265 (6)
H7	1.4707	0.4951	0.8803	0.032*
C8	1.3270 (2)	0.45372 (12)	0.7418 (2)	0.0197 (5)
H8	1.3642	0.4146	0.7337	0.024*
C9	1.18374 (19)	0.44264 (11)	0.3488 (2)	0.0157 (5)
C10	1.1833 (2)	0.51576 (12)	0.3709 (3)	0.0181 (5)
H10A	1.1011	0.5309	0.3492	0.022*
H10B	1.2273	0.5259	0.4686	0.022*
C11	1.2406 (2)	0.55027 (12)	0.2784 (3)	0.0190 (5)
H11	1.2398	0.5974	0.2943	0.023*
C12	1.1715 (2)	0.53586 (12)	0.1271 (3)	0.0226 (5)
H12A	1.2081	0.5583	0.0674	0.027*
H12B	1.0893	0.5512	0.1035	0.027*
C13	1.1723 (2)	0.46359 (13)	0.1033 (3)	0.0243 (6)
H13	1.1270	0.4539	0.0045	0.029*
C14	1.1149 (2)	0.42896 (13)	0.1953 (3)	0.0226 (5)
H14A	1.1140	0.3824	0.1782	0.027*
H14B	1.0322	0.4435	0.1723	0.027*
C15	1.3685 (2)	0.52770 (12)	0.3139 (3)	0.0211 (5)
H15A	1.4062	0.5504	0.2555	0.025*
H15B	1.4136	0.5376	0.4114	0.025*
C16	1.3704 (2)	0.45559 (12)	0.2898 (3)	0.0231 (5)
H16	1.4538	0.4408	0.3122	0.028*
C17	1.2999 (2)	0.44045 (13)	0.1382 (3)	0.0260 (6)
H17A	1.3369	0.4620	0.0777	0.031*
H17B	1.3008	0.3938	0.1224	0.031*
C18	1.3131 (2)	0.42047 (12)	0.3820 (3)	0.0209 (5)
H18A	1.3584	0.4293	0.4799	0.025*
H18B	1.3149	0.3738	0.3669	0.025*
C19	0.9811 (2)	0.26971 (11)	0.4941 (3)	0.0214 (5)
H19A	0.9957	0.2531	0.5882	0.026*
H19B	1.0165	0.2393	0.4457	0.026*
C20	0.8097 (2)	0.20785 (13)	0.3899 (3)	0.0288 (6)
H20A	0.8220	0.1840	0.4764	0.035*
H20B	0.8531	0.1856	0.3372	0.035*
C21	0.6788 (2)	0.21007 (15)	0.3057 (4)	0.0373 (7)
H21A	0.6671	0.2318	0.2169	0.045*
H21B	0.6476	0.1660	0.2859	0.045*
C22	0.6601 (2)	0.30815 (13)	0.4241 (3)	0.0260 (6)

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H22A	0.6190	0.3268	0.4836	0.031*
H22B	0.6444	0.3358	0.3420	0.031*
C23	0.7922 (2)	0.30593 (12)	0.5025 (3)	0.0244 (6)
H23A	0.8229	0.3500	0.5237	0.029*
H23B	0.8070	0.2831	0.5908	0.029*
C24	0.5325 (2)	0.21268 (12)	0.4181 (3)	0.0244 (6)
C25	0.3970 (3)	0.21595 (15)	0.5362 (4)	0.0396 (8)
H25A	0.4430	0.1885	0.6141	0.047*
H25B	0.3439	0.1883	0.4637	0.047*
C26	0.3270 (3)	0.26259 (15)	0.5833 (3)	0.0378 (7)
H26A	0.2717	0.2398	0.6178	0.057*
H26B	0.2829	0.2901	0.5061	0.057*
H26C	0.3800	0.2889	0.6572	0.057*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0274 (3)	0.0241 (3)	0.0246 (3)	-0.0048 (2)	0.0142 (3)	0.0009 (3)
N2	0.0183 (10)	0.0185 (10)	0.0267 (11)	-0.0019 (8)	0.0124 (9)	-0.0006 (9)
N1	0.0144 (9)	0.0165 (10)	0.0198 (10)	-0.0024 (7)	0.0087 (8)	-0.0013 (8)
N3	0.0194 (10)	0.0172 (10)	0.0254 (11)	-0.0037 (8)	0.0136 (9)	-0.0013 (9)
N4	0.0216 (10)	0.0192 (10)	0.0295 (12)	-0.0051 (8)	0.0139 (9)	-0.0057 (9)
N5	0.0229 (11)	0.0304 (12)	0.0428 (14)	-0.0083 (9)	0.0178 (10)	-0.0138 (11)
O1	0.0272 (10)	0.0253 (10)	0.0445 (12)	-0.0067 (8)	0.0206 (9)	-0.0046 (9)
O2	0.0336 (11)	0.0199 (10)	0.0443 (12)	-0.0039 (8)	0.0100 (9)	-0.0032 (9)
C1	0.0135 (11)	0.0154 (11)	0.0288 (13)	0.0002 (8)	0.0101 (10)	-0.0001 (10)
C2	0.0130 (10)	0.0156 (11)	0.0208 (12)	0.0002 (8)	0.0074 (9)	-0.0009 (9)
C3	0.0151 (11)	0.0177 (11)	0.0159 (11)	-0.0037 (9)	0.0068 (9)	0.0011 (9)
C4	0.0171 (11)	0.0222 (13)	0.0204 (12)	-0.0007 (9)	0.0073 (10)	-0.0004 (10)
C5	0.0312 (14)	0.0222 (13)	0.0238 (13)	-0.0026 (10)	0.0150 (11)	-0.0035 (11)
C6	0.0275 (13)	0.0327 (15)	0.0201 (13)	-0.0164 (11)	0.0098 (11)	-0.0058 (11)
C7	0.0152 (11)	0.0454 (17)	0.0182 (12)	-0.0065 (11)	0.0049 (10)	0.0013 (12)
C8	0.0182 (11)	0.0257 (13)	0.0168 (12)	0.0018 (9)	0.0081 (10)	0.0047 (10)
C9	0.0139 (10)	0.0168 (11)	0.0179 (11)	-0.0017 (8)	0.0074 (9)	-0.0009 (9)
C10	0.0187 (11)	0.0180 (12)	0.0198 (12)	-0.0007 (9)	0.0094 (10)	-0.0009 (10)
C11	0.0218 (12)	0.0166 (11)	0.0209 (12)	-0.0002 (9)	0.0103 (10)	0.0021 (10)
C12	0.0214 (12)	0.0278 (13)	0.0200 (12)	0.0019 (10)	0.0089 (10)	0.0055 (11)
C13	0.0263 (13)	0.0317 (14)	0.0170 (12)	-0.0070 (11)	0.0100 (11)	-0.0046 (11)
C14	0.0210 (12)	0.0280 (13)	0.0214 (13)	-0.0078 (10)	0.0108 (11)	-0.0052 (11)
C15	0.0189 (11)	0.0236 (13)	0.0225 (12)	-0.0055 (9)	0.0094 (10)	0.0013 (10)
C16	0.0185 (11)	0.0245 (13)	0.0308 (14)	0.0035 (10)	0.0144 (11)	0.0066 (11)
C17	0.0328 (14)	0.0221 (13)	0.0325 (15)	-0.0021 (11)	0.0234 (12)	-0.0031 (11)
C18	0.0158 (11)	0.0223 (12)	0.0277 (13)	0.0033 (9)	0.0114 (10)	0.0068 (11)
C19	0.0221 (12)	0.0147 (12)	0.0301 (14)	-0.0041 (9)	0.0124 (11)	-0.0018 (10)
C20	0.0298 (14)	0.0217 (13)	0.0408 (16)	-0.0058 (10)	0.0199 (13)	-0.0115 (12)
C21	0.0288 (15)	0.0368 (17)	0.0490 (19)	-0.0089 (12)	0.0165 (14)	-0.0236 (15)
C22	0.0241 (13)	0.0228 (13)	0.0357 (15)	-0.0023 (10)	0.0164 (12)	-0.0019 (12)
C23	0.0241 (13)	0.0194 (13)	0.0339 (15)	-0.0052 (10)	0.0152 (12)	-0.0085 (11)

C24	0.0193 (12)	0.0219 (13)	0.0295 (14)	-0.0001 (10)	0.0051 (11)	0.0001 (11)
C25	0.0350 (16)	0.0367 (17)	0.053 (2)	-0.0081 (13)	0.0231 (15)	0.0045 (15)
C26	0.0297 (15)	0.0428 (18)	0.0458 (18)	0.0003 (13)	0.0192 (14)	0.0069 (15)

Geometric parameters (Å, °)

S1—C1	1.664 (3)	C11—H11	1.0000
N2—C2	1.306 (3)	C12—C13	1.532 (4)
N2—N3	1.384 (3)	C12—H12A	0.9900
N1—C2	1.392 (3)	C12—H12B	0.9900
N1—C1	1.390 (3)	C13—C14	1.533 (3)
N1—C3	1.443 (3)	C13—C17	1.534 (4)
N3—C1	1.351 (3)	C13—H13	1.0000
N3—C19	1.453 (3)	C14—H14A	0.9900
N4—C19	1.457 (3)	C14—H14B	0.9900
N4—C23	1.467 (3)	C15—C16	1.530 (3)
N4—C20	1.468 (3)	C15—H15A	0.9900
N5—C24	1.350 (3)	C15—H15B	0.9900
N5—C22	1.453 (3)	C16—C17	1.539 (4)
N5—C21	1.457 (4)	C16—C18	1.539 (3)
O1—C24	1.348 (3)	C16—H16	1.0000
O1—C25	1.463 (3)	C17—H17A	0.9900
O2—C24	1.214 (3)	C17—H17B	0.9900
C2—C9	1.507 (3)	C18—H18A	0.9900
C3—C8	1.380 (3)	C18—H18B	0.9900
C3—C4	1.386 (3)	C19—H19A	0.9900
C4—C5	1.392 (4)	C19—H19B	0.9900
C4—H4	0.9500	C20—C21	1.522 (4)
C5—C6	1.390 (4)	C20—H20A	0.9900
C5—H5	0.9500	C20—H20B	0.9900
C6—C7	1.385 (4)	C21—H21A	0.9900
C6—H6	0.9500	C21—H21B	0.9900
C7—C8	1.383 (4)	C22—C23	1.521 (3)
C7—H7	0.9500	C22—H22A	0.9900
C8—H8	0.9500	C22—H22B	0.9900
C9—C10	1.547 (3)	C23—H23A	0.9900
C9—C14	1.546 (3)	C23—H23B	0.9900
C9—C18	1.548 (3)	C25—C26	1.476 (4)
C10—C11	1.536 (3)	C25—H25A	0.9900
C10—H10A	0.9900	C25—H25B	0.9900
C10—H10B	0.9900	C26—H26A	0.9800
C11—C12	1.530 (3)	C26—H26B	0.9800
C11—C15	1.532 (3)	C26—H26C	0.9800
C2—N2—N3	104.9 (2)	C9—C14—H14B	109.6
C2—N1—C1	108.66 (19)	H14A—C14—H14B	108.1
C2—N1—C3	129.67 (19)	C16—C15—C11	109.51 (19)
C1—N1—C3	121.6 (2)	C16—C15—H15A	109.8
C1—N3—N2	113.53 (19)	C11—C15—H15A	109.8
C1—N3—C19	126.2 (2)	C16—C15—H15B	109.8

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N2—N3—C19	119.4 (2)	C11—C15—H15B	109.8
C19—N4—C23	112.3 (2)	H15A—C15—H15B	108.2
C19—N4—C20	107.84 (19)	C15—C16—C17	109.6 (2)
C23—N4—C20	109.9 (2)	C15—C16—C18	109.6 (2)
C24—N5—C22	126.9 (2)	C17—C16—C18	108.8 (2)
C24—N5—C21	118.8 (2)	C15—C16—H16	109.6
C22—N5—C21	113.7 (2)	C17—C16—H16	109.6
C24—O1—C25	113.1 (2)	C18—C16—H16	109.6
N3—C1—N1	102.8 (2)	C13—C17—C16	109.7 (2)
N3—C1—S1	129.78 (19)	C13—C17—H17A	109.7
N1—C1—S1	127.46 (19)	C16—C17—H17A	109.7
N2—C2—N1	110.1 (2)	C13—C17—H17B	109.7
N2—C2—C9	122.7 (2)	C16—C17—H17B	109.7
N1—C2—C9	127.1 (2)	H17A—C17—H17B	108.2
C8—C3—C4	121.6 (2)	C16—C18—C9	110.43 (19)
C8—C3—N1	119.1 (2)	C16—C18—H18A	109.6
C4—C3—N1	119.3 (2)	C9—C18—H18A	109.6
C3—C4—C5	119.3 (2)	C16—C18—H18B	109.6
C3—C4—H4	120.4	C9—C18—H18B	109.6
C5—C4—H4	120.4	H18A—C18—H18B	108.1
C6—C5—C4	119.7 (2)	N3—C19—N4	111.9 (2)
C6—C5—H5	120.2	N3—C19—H19A	109.2
C4—C5—H5	120.2	N4—C19—H19A	109.2
C7—C6—C5	119.9 (2)	N3—C19—H19B	109.2
C7—C6—H6	120.1	N4—C19—H19B	109.2
C5—C6—H6	120.1	H19A—C19—H19B	107.9
C6—C7—C8	121.0 (2)	N4—C20—C21	109.2 (2)
C6—C7—H7	119.5	N4—C20—H20A	109.8
C8—C7—H7	119.5	C21—C20—H20A	109.8
C3—C8—C7	118.6 (2)	N4—C20—H20B	109.8
C3—C8—H8	120.7	C21—C20—H20B	109.8
C7—C8—H8	120.7	H20A—C20—H20B	108.3
C2—C9—C10	113.77 (19)	N5—C21—C20	110.0 (2)
C2—C9—C14	108.89 (19)	N5—C21—H21A	109.7
C10—C9—C14	107.90 (19)	C20—C21—H21A	109.7
C2—C9—C18	108.98 (19)	N5—C21—H21B	109.7
C10—C9—C18	108.48 (19)	C20—C21—H21B	109.7
C14—C9—C18	108.7 (2)	H21A—C21—H21B	108.2
C11—C10—C9	110.37 (19)	N5—C22—C23	110.5 (2)
C11—C10—H10A	109.6	N5—C22—H22A	109.6
C9—C10—H10A	109.6	C23—C22—H22A	109.6
C11—C10—H10B	109.6	N5—C22—H22B	109.6
C9—C10—H10B	109.6	C23—C22—H22B	109.6
H10A—C10—H10B	108.1	H22A—C22—H22B	108.1
C12—C11—C15	109.8 (2)	N4—C23—C22	110.8 (2)
C12—C11—C10	109.8 (2)	N4—C23—H23A	109.5
C15—C11—C10	109.5 (2)	C22—C23—H23A	109.5
C12—C11—H11	109.2	N4—C23—H23B	109.5
C15—C11—H11	109.2	C22—C23—H23B	109.5

C10—C11—H11	109.2	H23A—C23—H23B	108.1
C11—C12—C13	109.1 (2)	O2—C24—O1	123.5 (2)
C11—C12—H12A	109.9	O2—C24—N5	124.0 (3)
C13—C12—H12A	109.9	O1—C24—N5	112.4 (2)
C11—C12—H12B	109.9	O1—C25—C26	108.7 (2)
C13—C12—H12B	109.9	O1—C25—H25A	109.9
H12A—C12—H12B	108.3	C26—C25—H25A	109.9
C14—C13—C12	109.8 (2)	O1—C25—H25B	109.9
C14—C13—C17	109.4 (2)	C26—C25—H25B	109.9
C12—C13—C17	109.5 (2)	H25A—C25—H25B	108.3
C14—C13—H13	109.4	C25—C26—H26A	109.5
C12—C13—H13	109.4	C25—C26—H26B	109.5
C17—C13—H13	109.4	H26A—C26—H26B	109.5
C13—C14—C9	110.48 (19)	C25—C26—H26C	109.5
C13—C14—H14A	109.6	H26A—C26—H26C	109.5
C9—C14—H14A	109.6	H26B—C26—H26C	109.5
C13—C14—H14B	109.6		
C2—N2—N3—C1	-0.4 (3)	C11—C12—C13—C14	59.8 (3)
C2—N2—N3—C19	-170.7 (2)	C11—C12—C13—C17	-60.3 (3)
N2—N3—C1—N1	1.0 (3)	C12—C13—C14—C9	-60.4 (3)
C19—N3—C1—N1	170.5 (2)	C17—C13—C14—C9	59.8 (3)
N2—N3—C1—S1	-178.88 (18)	C2—C9—C14—C13	-177.0 (2)
C19—N3—C1—S1	-9.4 (4)	C10—C9—C14—C13	59.1 (3)
C2—N1—C1—N3	-1.2 (2)	C18—C9—C14—C13	-58.4 (3)
C3—N1—C1—N3	-179.45 (19)	C12—C11—C15—C16	-60.3 (3)
C2—N1—C1—S1	178.72 (18)	C10—C11—C15—C16	60.4 (3)
C3—N1—C1—S1	0.5 (3)	C11—C15—C16—C17	59.3 (3)
N3—N2—C2—N1	-0.4 (2)	C11—C15—C16—C18	-60.1 (3)
N3—N2—C2—C9	175.8 (2)	C14—C13—C17—C16	-60.6 (3)
C1—N1—C2—N2	1.0 (3)	C12—C13—C17—C16	59.7 (3)
C3—N1—C2—N2	179.1 (2)	C15—C16—C17—C13	-59.2 (3)
C1—N1—C2—C9	-175.0 (2)	C18—C16—C17—C13	60.6 (3)
C3—N1—C2—C9	3.1 (4)	C15—C16—C18—C9	59.7 (3)
C2—N1—C3—C8	-89.4 (3)	C17—C16—C18—C9	-60.1 (3)
C1—N1—C3—C8	88.4 (3)	C2—C9—C18—C16	177.3 (2)
C2—N1—C3—C4	91.4 (3)	C10—C9—C18—C16	-58.3 (3)
C1—N1—C3—C4	-90.7 (3)	C14—C9—C18—C16	58.8 (3)
C8—C3—C4—C5	1.3 (4)	C1—N3—C19—N4	110.8 (3)
N1—C3—C4—C5	-179.6 (2)	N2—N3—C19—N4	-80.2 (3)
C3—C4—C5—C6	-0.2 (4)	C23—N4—C19—N3	-71.6 (3)
C4—C5—C6—C7	-0.6 (4)	C20—N4—C19—N3	167.1 (2)
C5—C6—C7—C8	0.2 (4)	C19—N4—C20—C21	-176.4 (2)
C4—C3—C8—C7	-1.7 (4)	C23—N4—C20—C21	60.9 (3)
N1—C3—C8—C7	179.2 (2)	C24—N5—C21—C20	-117.4 (3)
C6—C7—C8—C3	0.9 (4)	C22—N5—C21—C20	55.1 (3)
N2—C2—C9—C10	141.9 (2)	N4—C20—C21—N5	-58.0 (3)
N1—C2—C9—C10	-42.6 (3)	C24—N5—C22—C23	118.9 (3)
N2—C2—C9—C14	21.5 (3)	C21—N5—C22—C23	-52.9 (3)
N1—C2—C9—C14	-163.0 (2)	C19—N4—C23—C22	-179.4 (2)

supplementary materials

N2—C2—C9—C18	-96.9 (3)	C20—N4—C23—C22	-59.3 (3)
N1—C2—C9—C18	78.6 (3)	N5—C22—C23—N4	54.2 (3)
C2—C9—C10—C11	-179.92 (19)	C25—O1—C24—O2	7.5 (4)
C14—C9—C10—C11	-59.0 (2)	C25—O1—C24—N5	-173.4 (2)
C18—C9—C10—C11	58.6 (2)	C22—N5—C24—O2	-169.5 (3)
C9—C10—C11—C12	60.4 (2)	C21—N5—C24—O2	1.9 (4)
C9—C10—C11—C15	-60.2 (3)	C22—N5—C24—O1	11.4 (4)
C15—C11—C12—C13	60.6 (3)	C21—N5—C24—O1	-177.2 (2)
C10—C11—C12—C13	-59.9 (3)	C24—O1—C25—C26	-171.0 (2)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C6—H6 \cdots O2 ⁱ	0.95	2.48	3.406 (3)	166
C8—H8 \cdots O2 ⁱⁱ	0.95	2.46	3.207 (3)	135

Symmetry codes: (i) $-x+2, y+1/2, -z+3/2$; (ii) $x+1, -y+1/2, z+1/2$.

Fig. 1

